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Chapter 19 Ionic Equilibria In Aqueous Systems  
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\_\_\_\_\_ 1. Which Of The Following Aqueous Mixtures Would Be A Buffer System? A. HCl, NaCl B. HNO<sub>3</sub>, NaNO<sub>3</sub> C. H<sub>3</sub>PO<sub>4</sub>, H<sub>2</sub>PO<sub>4</sub><sup>-</sup> D. H<sub>2</sub>SO<sub>4</sub>, CH<sub>3</sub>COOH E. NH<sub>3</sub>, NaOH  
2. Which, If Any, Of The Following Aqueous Mixtures Would Be A Buffer System?  
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