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### **NAMING ACIDS Name The Following Acids. Name 2. 3. 4. 5. 6 ...**

NAMING ACIDS Name The Following Acids. Name 2. 3. 4. 5. 6. 8. 9. 10, HNO<sub>3</sub> HCl H<sub>2</sub>SO<sub>4</sub> HBr H<sub>2</sub>CO<sub>3</sub> Act Acid Nom [ Acùd Write The Formulas Of The Following Acids. 2th, 2024

### **#82 Notes Unit 10: Acids & Bases Ch. Acids, Bases, And ...**

#82 Notes Unit 10: Acids & Bases Ch. Acids, Bases, And Solubility Ch. Acids/ Bases \*\* Acids: Start With "H" Or End In -CO<sub>2</sub>H (-COOH) And Have A Sour Taste. \*\* Bases: End In -OH (positive Ion In Front) Or End In -NH<sub>2</sub>, -NH<sub>3</sub><sup>+</sup>. 2, NH<sub>3</sub>. 3. They Have A Bitter Taste And Feel Slippery. \* 3 Acid/Base Theories . I. Arrhenius Theory A) Acids Produce H<sup>+</sup> 2th, 2024

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1. NAMING BINARY ACIDS: The Name Of The Binary Acid Consists Of Two Words. The First Word Has Three Parts: The "hydro" Prefix The Root Of The Nonmetal Element The "ic" Ending The Second Word Is Always "acid" Examples: HCl = Hydro Chlor Ic Acid = Hydrochloric Acid HBr = Hydro Brom Ic Acid = Hydrobromic Acid 1th, 2024

### **Naming Acids Worksheet Key Name The Following Acids**

Naming Acids Worksheet Key Block\_\_\_\_ Name\_\_\_\_ Name The Following Acids 1. HCL Hydrochloric Acid 2. HClO4 Perchloric Acid 3. HIO3 Iodic Acid 4. HI Hydroiodic Acid 5. H2SO4 Sulfuric Acid 6. H2S Hydrosulfuric Acid 7. HCN Hydrocyanic Acid 8. H2CO3 Carbonic Acid 9. 2th, 2024

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