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Differential Rate Laws Vs. Integrated Rate Laws Important Terms And Equations For Radioactive Decay Terms  $N_t$  = Number Of Radioactive Atoms At Some Elapsed Time,  $T$   $N_0$  = Number Of Radioactive Atoms At An Initial Point, At Time  $T = 0$   $k$  = Decay Constant In Units Of  $(\text{time})^{-1}$  (analogous To  $K$ )  $A$  = Decay Activity In Units Of Disintegrations Per Time Per Gram (analogous To Rate)  $T_{1/2}$  = Half-life Period; Time Required For  $N_t$  To Fall To May 3th, 2024

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