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1. Describe How Gases, Liquids, And Solids Compare Using The Following Table. Solids Liquids Gases Volume (definite Or Indefinite) Molecular Motion (high, Med, Low) Distance Between Molecules (g 2. Read Page 420: What Is Kinetic Energy? Describe Kinetic Molecular ... 1th, 2024

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### **CHAPTER 12 GASES AND KINETIC-MOLECULAR THEORY**

3 Boyle's Law: The Volume-Pressure Relationship  $\{V \propto 1/P$  Or  $\{V = K (1/P)$  Or  $PV = K\{P_1V_1 = K_1$  For One Sample Of A Gas.  $\{P_2V_2 = K_2$  For A Second Sample Of A Gas.  $\{k_1 = K_2$  For The Same Sample Of A Gas At The Same T.  $\{$ Thus We Can Write Boyle's Law Mathematically As  $P_1V_1 = P_2V_2$  4th, 2024

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A) Degrees Celsius Are Larger Than Degrees Kelvin. B)  $0^{\circ}\text{K}$  Is Much Warmer Than  $0^{\circ}\text{C}$ . C) Celsius Is A Relative Scale Is Based On The Freezing And Boiling Of Water; Kelvin Is An Absolute Scale. D) There Is No Difference whatsoever Between Them. 20. What Is The Difference Between The 4th, 2024

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