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| | | | | | | | |
|----------------------------|-----------------------------|---------------------|--------------------------|--------|--------------------|------|------|
| AlF ₃ Na | Sodium Tetrafluoroaluminate | -1869.0 | -1827.5 | 345.7 | 105.9 | | |
| AlH ₃ | Aluminum Hydride | 259.2 | 231.2 | 187.9 | 29.4 | | |
| AlH ₃ | Aluminum Hydride | -46.0 | 30.0 | 40.2 | AlH ₄ K | | |
| Potassium Aluminum Hydride | -183.7 | AlH ₄ Li | Lithium Aluminum Hydride | -116.3 | -44.7 | 78.7 | 83.2 |
| AlH ₄ Na | Sodium Aluminum Hydride | -15.5 | All Aluminum Monoiodide | 65.5 | 36.0 | | |

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Webbook, Molecular Formula Crystal Liquid Gas Name F H ° KJ/mol F G ° KJ/mol S °
J/mol K C P J/mol K F H ° KJ/mol F G ° KJ/mol S ° J/mol K C P J/mol K F H ° KJ/mol F G °
KJ/mol S ° J/mol K C P J/mol K Substances Not Containing Carbon: Ac Actinium 0.0
56.5 27.2 406.0 366.0 188.1 20.8 Ag Silver 0.0 42.6 ... Jun 13th,
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Same As Enthalpycalculations. There Are Tables Of Standard Free Energies Of
Formation Of Compounds.Elements In Their Standard States Areassigned
Zeroastheir Gibbs Free Energy Of Formation. Consider The Following Chemical
Reaction: $\text{H}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightarrow 2\text{HCl}(\text{g})$. Jun 20th, 2024.

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Formation And Energy Of HOMO. The Second PC Is Strongly Influenced By Energy Of
LUMO, While Substitution Pattern Parameters, Number Of Chlorine Atoms At α -
positions And Vicinal (adjac-ent) Carbon Atoms Jan 23th, 2024Selected Values Of
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Valueof ΔH_f° represents theapparent Enthalpy Ofthe Reaction Of Formation Ofthe
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