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## Intro To Energy Model PhET Lab Name Hr

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8. On The Back Of This Worksheet, Determine The Average Atomic Mass Of Your Element Using This Equation: Be Sure To Show Your Work! Average Atomic Mass = (avg. Mass Of Isotope)(%) + (avg. Mass Of Isotope)(%) + ... (Remember That % Means "pe Mar 2th, 2024

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(4) 238U (t1/2 = 4.5 X 109 Y) Decays To 234Th (t1/2 = 24.1 D). Define And Give The Conditions For Secular Equilibrium. Assuming The Decay Of 238U = 2.3 Dpm Kg-1 What Is The Molar Concentration Of 234Th At Secular Equilibrium. Assume That Here Has Been An Extreme Bloom Period With Intensive Scavenging That Removes Most Of The 234Th Jul 2th, 2024

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