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Thermochemistry183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505-510) This Section Explains The Relationship Between Energy And Heat, And Distinguishes Between Heat Capacity And Specific Heat. 1th, 2024SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORKSECTION 17.2 MEASURING AND EXPRESSING ENTHALPY CHANGES 1. A Student Mixed 75.0 ML Of Water Containing 0.75 Mol HCl At 25 C With 75.0 ML Of Water Containing 0.75 Mol Of NaOH At 25 C In A Foam Cup Calorimeter. The Temperature Of The Resulting Solution Increased To 35 C. How Much Heat In Kilojoules Was Released By This Reaction? C Water 4.18 //gp C 2. 3th, 2024. Flow Of Energy Heat And Work AnswersThermochemistry 17 1- The Flow Of Energy Heat And Work Exothermic Endothermic Calorie Joule

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