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Free-energy Change (  $\Delta G$  ) And Entropy Change (  $\Delta S$  ) - Chemrevise Nov 03, 2018 · Gibbs Free Energy Change,  $\Delta G$  Gibbs Free Energy Is A Term  
That Combines The Effect Of Enthalpy And Entropy Into One Number The Balance Between Entropy And Enthalpy  
Determines The Feasibility Of A Reaction. This Is Given By The Relationship :  $\Delta G = \Delta H - T\Delta S_{\text{system}}$  For Any Spontaneous  
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