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N0O +ne  $\$  NRR (2.1.36) The Free Energy Of The Cell Reaction Is Given By, DG = DGO + RT Ln AnR R A N H+ H AnO O A NH 2 H2 (2.1.38) Which Since DG = NFErxn And A H+ = AH 2 = 1 Becomes, E = EO + RT NF Ln AnR R AnO O. (2.1.40) This Expression Is Called The Nernst Equation. This Relation Can 2th, 2024

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