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Differential Rate Laws Vs. Integrated Rate Laws

Important Terms And Equations For Radioactive Decay

$N_t = N_0 e^{-\lambda t}$

N_t = Number Of Radioactive Atoms At Some Elapsed Time, T

N_0 = Number Of Radioactive Atoms At An Initial Point, At Time $T = 0$

λ = Decay Constant In Units Of (time)⁻¹ (analogous To K)

A = Decay Activity In Units Of Disintegrations Per Time Per Gram (analogous To Rate)

$T_{1/2}$ = Half-life Period; Time Required For N_t To Fall To

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