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0.50 g of the carbon combusts to form CO. The resultant mixture of CO₂ and CO is passed through excess NaOH. Answer on Question #58182 Chemistry Organic Chemistry ... 2) Aniline and acetanilide stand in a one-to-one molar relationship, so 0.0322 mol of aniline produces 0.0322 mol of acetanilide. The amount of acetic anhydride does not play a role, since it is in excess. Then, n(CO) = 0.0322 mol. Aniline: n(CO) = 0.0322 mol. We find the molar mass of acetanilide: (C₈H₉NO) = 135.13 g/mol. 89 mg acetanilide ... Jun 24th, 2024 Unit 13: Organic Chemistry-Key Regents Chemistry '14 Mr ... Unit 13: Organic Chemistry-Key Regents Chemistry '14-'15 Mr. Murdoch Page 12 of 65 Website Upload 2015 Organic Chemistry Lecture Key Reference Table Q, Homologous Series of Hydrocarbons, Has the suffix information, formula, naming, and structural characteristics. It was already Mar 13th, 2024.

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