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A) [Co(H₂O)₆]²⁺ + Co + 2 e) [FeO₄]²⁻ - Fe + 6 l) K₂[CoCl₄] + Co + 2 B) [CrCl₆]³⁻ - Cr + 3 F) [Mn(CN)₆]⁴⁻ - Mn + 2 J) K₃[AuF₆] + Au + 3 C) [Co(NH₃)₆]Cl₂ + Co + 2 G) [Ni(CO)₄] + Ni O K) (NH₄)₂[IrCl₆] + Ir + 4 D) [Co(NH₃)₅Cl]Cl₂ + Co + 3 H) [Ni(EDTA)]²⁻ - Ni + 2 L) Na[Mn(CO)₅] + Mn - 1 TASK 2 - Writing Half Equations A) 2I⁻ → I₂ + 2e⁻ B ... 5th, 2024

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Metal Carbonate → Metal Oxide + Carbon Dioxide (on Heating) CaCO₃ → CaO + CO₂ TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. 17th, 2024

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