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Chemistry 11 - Balancing Chemical Equations 1. Balance The Following Equations A. $\text{NH}_3 + \text{O}_2 \rightarrow \text{NO} + \text{H}_2\text{O}$ B. $(\text{NH}_4)_2\text{C}_2\text{O}_4 + \text{AlCl}_3 \rightarrow \text{Al}_2(\text{C}_2\text{O}_4)_3 + \text{NH}_4\text{Cl}$ C. $\text{C}_{13}\text{H}_{28} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ D. $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2$ F 2th, 2024

Writing And Balancing Chemical Equations Chemistry 110

Writing And Balancing Chemical Equations Chemistry 110 I] On The Line At The Left, Write The Letter Corresponding To The Reaction Type: (A) Combination/Synthesis (B) Decomposition (C) Replacement/single Replacement 2th, 2024

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