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Born-Haber Cycle For BaO Crystal. Step 1. Sublimation

Of Solid Ba:  $\text{Ba(s)} \rightarrow \text{Ba(g)}$ ,  $\Delta H_1 = 180.0 \text{ KJ/mol}$  During This Process, The Change In Enthalpy Is Positive, Energy Is Required To Change The State Of Barium From Solid To Gaseous (to Break Metal Lattice). Step 2. May 1th, 2024  
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