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Chemical Quantities 1-carbon 2-oxygen Thus One Mole Of  $\text{CO}_2$  Contains Three Times Avogadro's # Of Atoms. Example: How Many Atoms Are In 2.12 Mole Of Propane ( $\text{C}_3\text{H}_8$ )? Know: # Of Moles = 2.12 Mole  $\text{C}_3\text{H}_8$  1 Mole  $\text{C}_3\text{H}_8$  =  $6.02 \times 10^{23}$  Molecules  $\text{C}_3\text{H}_8$  1 Molecule  $\text{C}_3\text{H}_8$  = 11 Atoms ( 3 Carbon Atoms & 8 Hydrogen Atoms) Feb 2th, 2024

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