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Example 9-4. Absorption Of SO 2 The Molar Density Of The Feed Gas May Be Determined From The Formula (mol/L)=12.2P(atm)T-1(K) Where P And T Are The Absolute Pressure And Temperature Of The Gas. The Following Data Are Taken For The Feed Gas: T=750F, P=150 Psig, H=210 Mm, R=82.4 And For The Outlet Gas, R=11.6. Determine The (a) Mass Fraction Of SO 1th, 2024

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That Comes From Each Of The Elements In The Compound By Using This Equation: % Element = (number Of Atoms)(atomic Weight) (FW Of The Compound) X 100. 3th, 2024

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Stoichiometry Calculating Empirical Formulas The Compound Para-aminobenzoic Acid (you May Have Seen It Listed As PABA On Your Bottle Of Sunscreen) Is Composed Of Carbon (61.31%), Hydrogen (5.14%), Nitrogen (10.21%), And Oxygen (23.33%). Find The Empirical Formula Of PABA. 1th, 2024

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