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Chapter 3 Acid-Base Equilibria Acid-Base Equilibria Acids And Bases Play A Key Role In A Number Of Environmentally Important Chemical Reactions, Including Weathering, Transport Of Metals In Solution, And CO 2 Atmosphere-water Equilibria. In This Chapter We Will Develop The Concept Of An Acid And A Base, Characterize Strong And Weak Acids, Mar 2th, 2024

(aq) Acid Base Conjugate Conjugate Acid Base

Acid Base Conjugate Conjugate . Acid Base . 2) What Is The Strongest Base In The Following Reaction? HNO3(aq) + H. 2. O(I) Æ NO3-(aq) + H3O + (aq) H2O Is The Strongest Base. Strong Acids, Such As HNO3 Have Weak Conjugate Bases, So NO3-is A Weak Base. H 2O And May 17th, 2024

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Carboxylic Acids Protonated Ketone-7.3 6.37 7 Carbonic Acid Tosic Acid -0.6 Protonated Pyridine 5.2 PKa Chart Conjugate Acid Conjugate Base Conjugate Acid Conjugate Base S T R O N G E S T A C I D S W E A K E S T B A S E S Hydrogen Sulfi Mar 15th, 2024

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