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Balancing Chemical Equations Worksheet 1 - ANSWERS

1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2. $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3. $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$ 4. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ 5. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ + O_2 6. $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$ 7. $2\text{Li} + \text{F}_2 \rightarrow 2\text{LiF}$ 8. $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$ 9. $2\text{NH}_3 \rightarrow 3\text{N}_2 + 2\text{H}_2$ 10.

$2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$... Jul 3th, 2024Balancing Chemical Equations - Answer Key

Balancing Chemical Equations - Answer Key Balance The Equations Below: 1) $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ 3) $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$ 4) $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 5) $1\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$ 6) $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$ 7) $1\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O}$

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