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To The Root Mean Squared Velocity, V. Rms, By L The Mean Kinetic Energy Of A Sample Of Ideal Gas Is Directly Proportional To Absolute Temperature. L The Mean Ki 13th, 2024CHAPTER 12 GASES AND KINETIC-MOLECULAR THEORY3 Boyle's Law: The Volume-Pressure Relationship  $\{V \propto 1/P \text{ Or} \{V = K (1/P) \text{ Or } PV = K\{P1V \ 1 = K \ 1 \text{ For One Sample Of A Gas. } \{P2V \ 2 = K \ 2 \text{ For A Second Sample Of A Gas. } \{k1 = K \ 2 \text{ For The Same Sample Of A Gas At The Same T. } \{Thus We Can Write Boyle's Law Mathematically As P 1V 1 = P 2V 2 4th. 2024.$ 

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 $26.7^{\circ}\text{C}$  . Physics 20: Heat Teacher Notes Page 8 Of 24 Practice Problems K = C + 273 C = K - 273 1. Convert Th 16th, 2024.

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